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Chemistry Chapter 8 Covalent Bonding

1. In covalent bonds, electron sharing

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usually occurs so that atoms attain the electron configuration of noble gases. (8) 2.

Atoms form double or triple covalent bonds if they can attain a noble gas structure by sharing two pairs or three pairs of electrons.

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Polar Covalent

Nonpolar Covalent

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Test. PLAY. Match.

Gravity. Created by.

brennallheart. Terms in

this set (59) Covalent

bond. a bond formed

by the sharing of

electrons between

atoms. Molecule. a

neutral group of atoms

joined together by

covalent bonds.

Diatomic molecule.

**Chemistry Chapter
8: Covalent Bonding
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Covalent Bonding.

Covalent bond.

covalent. molecule. the

same. atoms held

together by sharing

electrons. in a ... bond,

atoms take place in a

"tug of war" over

electro.... a neutral

group of atoms joined

together by covalent

bonds. all molecules in

a sample of a singular

molecular compound

are....

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covalent bonding

Flashcards and

Study ...

A covalent bond in which the bonding electrons are most likely to be found in sausage-shaped regions above and below the bond axis of the bonded atoms

Polar molecule A molecule in which one side of the molecule is slightly negative and the opposite side is slightly positive.

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Technically, any covalent bond between two different elements is polar. However, the degree of polarity is important. A covalent bond between two different elements may be so slightly imbalanced that the bond is, essentially, nonpolar. A bond may be so polar that an

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electron actually

transfers from one

atom to another,

forming a true ionic

bond.

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- Introduction to ...**

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8.1 The Covalent Bond

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pages 240-247

Practice Problems page

244 Draw the Lewis
structure for each

molecule. 1. PH₃ H₂ H₂

H— H H P respectively,

for single, double, and

triple P — — 2. H₂ S H

H H — H S S ...

Covalent

Bonding Covalent

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Answers gas structure

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by sharing two pairs or
three pairs of

electrons. Chemistry

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A covalent bond in which three pairs of electrons are shared by two atoms.

Answers

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Bonding Work 242

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Bonding Single

Covalent Bonds When only one pair of electrons is shared, such as in a hydrogen molecule, it is a single covalent bond. The

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shared electron pair is often referred to as the bonding pair.

Chapter 8 Covalent Bonding Work With Answers

molecular orbital that lies along the axis connecting two atom.... coordinate covalent bond. a bond in which one atom contributes both bonding electrons to.... bond dissociation energy. the total energy required to

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break the bond

between two covalen...

42 terms.

connorhumphrey2.

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Covalent Bonds.

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Valence bond theory

describes a covalent

bond as the overlap of

half-filled atomic

orbitals (each

containing a single

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electron) that yield a pair of electrons shared between the two bonded atoms. We say that orbitals on two different atoms overlap when a portion of one orbital and a portion of a second orbital occupy the same region of space.

8.1 Valence Bond

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Bonding. 20 terms.

Owen_Chandler3.

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terms. gregjones12.

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4.7: Organic Chemistry

Organic chemistry is

the study of the

chemistry of carbon

compounds. Organic

molecules can be

classified according to

the types of elements

and bonds in the

molecules. 4.8:

Covalent Bonding and

Simple Molecular

Compounds (Exercises)

These are homework

exercises to

accompany Chapter 4

of the Ball et al. "The

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Basics of GOB ...

Covalent Bonding

4: Covalent Bonding and Simple

Molecular

Compounds ...

Covalent bonds form when electrons are shared between atoms and are attracted by the nuclei of both atoms. In pure covalent bonds, the electrons are shared equally. In polar covalent bonds, the electrons are shared unequally, as

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one atom exerts a

stronger force of

attraction on the

electrons than the

other.

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