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126 Chemistry: Matter and Change •
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265-270 Section 8.5 Assessment page
270 68. Summarize how
electronegativity difference is related to
bond character. The greater the
electronegativity difference, the greater
the ionic nature of the bond. 69.

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Chapter 8 - Covalent Bonding - 8
Assessment - Page 257: 68 Answer

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Boiling points require the molecules to separate , when the state changes from a liquid into a gas.

Chemistry (12th Edition) Chapter 8 - Covalent Bonding - 8 ...

210 Chapter 8 Ionic Compounds

CHAPTER 8 What You'll Learn You will define a chemical bond. You will

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describe how ions form. You will identify ionic bonding and the characteristics of ionic compounds. You will name and write formulas for ionic compounds. You will relate metallic bonds to the characteristics of metals. Why It's Important

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Chapter 8 Assessment Chemistry Answers

Section 8.1 Assessment page 247 7.
Identify the type of atom that generally forms covalent bonds. The majority of covalent bonds form between nonmetallic elements. 8. Describe how the octet rule applies to covalent bonds. Atoms share valence electrons; the

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shared electrons complete the octet of each atom. 9. Illustrate the formation of single, double, and

Covalent Bonding

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(g) $\rightarrow 2\text{Li}_3\text{N}(s)$ If 5.00 g of each reactant
undergoes a reaction How I Got an 800
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inRegister. Hide.

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within a single molecule) It commonly bonds to unshared electron pairs on another electronegative atom.

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